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Note

Solid-state fluorodensitometric quantitation of aryInaphthalene lignan lactones of *Cleistanthus collinus*

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Arylnaphthalene lignan lactones are naturally occurring dimers in plants and known to have potential use in medicine^{1,2}. A few of them are reported to be highly toxic; others find use as antioxidants and insecticides³.

Cleistanthus collinus (Roxb.) Benth. & Hook.f. (family: Euphorbiaceae) is a highly poisonous plant. All parts of it are reported to be toxic and employed as a suicidal, homicidal, cattle and fish poison and for procuring criminal abortion⁴⁻⁶. Chemical characterization of different parts of the plant has led to the isolation and identification of certain lignan lactones including diphyllin, cleistanthin A, cleistanthin B and collinusin^{7,8}. Although the poisonous nature of the plant has been established⁹, the compound(s) responsible for the toxicity is not known. Nevertheless, few reports on the medicinal properties of the lignan lactones have appeared^{2,10-12}.

While information on the structures and biological activities of the constituents is available, there are little data on the quantitative analytical aspects. Methods reported in the literature cover only the qualitative identification of C. collinus from poisoned visceral materials and chromatographic systems for the separation of the lignan lactones^{13,14}.

Based on the photometric properties (including fluorescence under UV irradiation) of the four compounds mentioned above, we previously developed spectrophotometric and spectrofluorometric methods^{15,16}. The present paper deals with an improved technique, namely, the solid-state fluorodensitometric quantitation of diphyllin, cleistanthin A, B and collinusin —the major arylnaphthalene lignan lactones of *C. collinus*— after their thin-layer chromatographic (TLC) separation. While the earlier methods require TLC and subsequent concentration of the compounds, the present method is a direct one. The variation in the fluorescence characteristics of the compounds in solution and on silica gel (TLC) has also been studied and is utilized to enable a better quantitation.

EXPERIMENTAL

Instrumentation

All densitometric measurements were made with a soft laser (SL-TRFF) scanning densitometer (Biomed., U.S.A.).

Reagents

Chemicals and solvents were of analytical/spectral grade quality.

Authentic samples of diphyllin, cleistanthin A and collinusin were obtained from Ciba-Geigy (Bombay, India) and cleistanthin B from Osmania University, Hyderabad, India. The purity of the samples was verified from their spectral and chromatographic data.

Standard solutions of 500 μ g/ml were prepared in ethanol and working standard solutions were obtained by suitable dilution as and when needed.

Quantitative TLC

Known volumes of authentic sample solutions of diphyllin, cleistanthin A, B and collinusin were spotted on a TLC plate (Kieselgel 60G, E. Merck, 0.25 mm, activated at 110°C for 30 min) at a distance of 4 cm from the bottom with a microlitre syringe. The plate was developed to a distance of 15 cm with *n*-heptanechloroform-ethanol (50:50:5). It was then dried and viewed under UV light to ensure the resolution of the compounds.

Densitometric quantitation

The UV lamp of the densitometer was allowed to warm up for 15 min prior to use to obtain the best results. The plate was mounted upright on the stage with suitable spacers on the top and bottom of the plate. A low-pressure mercury lamp with an emission peak at 366 nm was used to excite the sample. A cobalt blue filter forms an integral part of the lamp and was used to block emission above 400 nm. The source was placed on the same side of the sample as the detector. The light (fluorescence) emitted from the sample was passed through a collimating slit. A secondary filter that blocks energy below 410 nm was placed between the slit and the detector to prevent UV light from the exciting source from reaching the detector. Thus, only light emitted from the sample is transmitted to the detection and measuring systems. The photomultiplier tube voltage, the analog gain and zero suppression control knobs were adjusted to give 85–90% full-scale deflection when the densest spot was scanned. The scan length was adjusted to a fixed distance by means of a thumb screw provided in the slot of the stage carriage.

The densitometer is equipped with a time-based electronic integrator which relates position and time in the quantitation of the sample, and was used to quantitate the area under the curve.

Calibration curve

The linearity between the fluorescence intensity and concentration was established by constructing calibration curves for all the four compounds.

Dried leaves of C. collinus were extracted with alcohol after defatting with hexane as reported earlier¹⁶. An aliquot of the final extract was chromatographed and subjected to densitometric measurements.

Recovery after spiking

Known amounts of the authentic samples (5 μ g in ethanol) of the lignan lactones were added to blood and tissue homogenate (prepared by homogenization of 1 g rat stomach with 1 ml of water) and processed for densitometric quantitation following the procedure reported elsewhere¹⁶.

RESULTS AND DISCUSSION

The chemical structures of the four compounds studied are shown in Fig. 1. The chromatographic resolution of the four compounds along with a leaf extract of *C. collinus* is shown in Fig. 2. Typical densitograms of the mixture of authentic samples of the four compounds and that of the leaf extract are given in Figs. 3 and 4, respectively. As shown, the leaf extract contains a few more fluorescent compounds apart from the lignans, the identification of which remains to be established.

COMPOUND	STRUCTURE	R _F VALUE*
CLEISTANTHIN A	$R = 3, 4 \text{ di} - 0 - \text{methyl}$ $x \text{ ylose}$ OR $H_{3}CO$ $H_{3}CO$ $H_{3}CO$ I	0.37
CLEISTANTHIN B	$ \begin{array}{c} $	0-01
DIPHYLLIN		0.22
COLLINUSIN		O·55

Fig. 1. Chemical structures and R_F values of the lignan lactones from the leaves of C. collinus. TLC was carried out on Kieselgel 60G with *n*-heptane-chloroform-ethanol (50:50:5) as mobile phase.



Fig. 2. UV photograph of the TLC resolution of C. collinus lignan lactones: from left to right, cleistanthin A (R_F 0.37), B (R_F 0.01), leaf extracts of C. collinus (collected from two different places separated by a distance of 200 km), collinus (R_F 0.55) and diphyllin (R_F 0.27).

Calibration graphs were constructed for all the four compounds individually by plotting the percentage peak area against concentration. The linearity extends over the range of 0.25–1.50 μ g for diphyllin, cleistanthin A and collinusin and 0.10–1.0 μ g for cleistanthin B (Fig. 5).

Intensity of fluorescence with time

In order to monitor the change, if any, in the intensity of fluorescence with time, the compounds after chromatography were subjected to densitometric evaluation at intervals over a period of 24 h. The intensity of fluorescence of cleistanthin A, B and collinus remains constant, while that of diphyllin decreases after 12 h, the effect being predominant at low concentrations.

Sensitivity

The lowest limit of sensitivity of the proposed method is 0.25 μ g of diphyllin, cleistanthin A and collinusin and 0.10 μ g of cleistanthin B. A comparison of sensitivity obtainable by the above method and that by spectrofluorometry¹⁶ is provided in Table I.

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Fig. 3. Densitogram of a mixture of four lignan lactones (each 1 μ g in ethanol) of C. colinus after TLC. Peaks (left to right): cleistanthin B, diphyllin, cleistanthin A and collinusin.

TABLE I

ARYLNAPHTHALENE LIGNAN LACTONES OF C. COLLINUS: COMPARISON OF SENSITIV-ITY OBTAINED IN SPECTROFLUOROMETRY AND FLUORODENSITOMETRY

Compound	Spectrofluorometry* (µg/ml)	Fluorodensitometry (µg)
 Diphyllin	0.1	0.25
Cleistanthin A	1	0.25
Cleistanthin B	1	0.10
Collinusin	**	0.25

* Minimum of 3 ml solution (0.3, 3, 3 μ g) is required for measurement.

** No response even at a concentration of 600 $\mu g/3$ ml in ethanol.



Fig. 4. Densitogram of C. collinus leaf extract¹⁶ after TLC. Peaks: 1 = cleistanthin B; 3 = diphyllin; 5 = cleistanthin A; 6 = collinusin.

Reproducibility

The precision of the method using densitometric analysis was evaluated by carrying out the experiments with a sample containing all the four lignan lactones in

Compound	Arnount	Percentas	ge peak area	_							Mean	S.D.	C.V.
	present (µg)	1	2	n	4	5	6	2	80	6			
Diabullin	0.50	11.875	11.744	11.736	10.528	12.347	11.354	10.515	11.355	10.310	11.29	0.56	5.03
Claictanthin A	1 25	35.608	37.072	34.576	36.623	36.267	35.281	35.43	36.706	37.548	36.123	0.97	2.68
Cleistanthin R	0.50	12.316	13.438	12.417	13.341	12.586	12.146	12.162	12.248	12.571	12.56	0.502	3.99
Collinusin	0.25	5.835	6.480	6.826	6.392	6.415	5.828	5.704	6.415	5.725	6.18	0.41	6.65

TABLE II



Fig. 5. Fluorodensitometric calibration curves for lignans from C. collinus: $\times - \times$, diphyllin; $\bullet - \bullet$ cleistanthin A; $\bigcirc - \bigcirc$, cleistanthin B; $\nabla - \bigtriangledown$, collinusin.

known quantity over a period of 9 days. The results presented in Table II indicate a precision of measurement of about $\pm 2.68-6.65\%$ as determined by the coefficient of variation (c.v.) for the four compounds. However, to obtain better results it is suggested that a calibration curve be constructed for each plate.

Recovery data from spiked biological samples of blood and tissue homogenate are given in Table III.

An interesting observation during the course of our investigation is that collinusin (a structural analogue of diphyllin), which could not be estimated by spectrofluorometry even at a concentration of $600 \ \mu g/3$ ml, was quantitatively analysed by the present fluorodensitometric method at a concentration of 0.25 μg ; also the intensity of luminescence on chromatograms remains unaltered for more than a fortnight. It is also interesting to note that diphyllin, which offered the maximum spectrofluorometric turn-out with unaltered luminescence in solution over a number of days, occupies the last place in the series when scanned for fluorodensitometry (Table I). As shown in Fig. 3, the relative order of solid state fluorescence intensity is cleistanthin B > A > collinusin > diphyllin, whereas in spectrofluorometry, it is di-

TABLE III

Specimen	Diphyllin		Cleistanti	hin A	Cleistanthin B		Collinusin	
	Amount added (µg)	% Recovery	Amount added (µg)	% Recovery	Amount added (µg)	% Recovery	Amount added (µg)	% Recovery
Blood	5	93.7	5	94.5	5	92.8	5	92.7
Tissue homogenate	5	90.3	5	90.7	5	88.5	5	88.1

RECOVERY OF LIGNAN LACTONES OF C. COLLINUS FROM SPIKED RAT STOMACH HOMOGENATE.

phyllin > cleistanthin B > cleistanthin A. Similar observations have been reported for aflatoxins B_1 , B_2 , G_1 and G_2^{17} .

CONCLUSIONS

Many reports on the medicinal utility of lignan lactones have appeared in the literature. Compounds like podophyllotoxin, α - and β -peltatin have been found to be active against tumours³. Similarly, diphyllin which occurs also in Diphyllia gravii and Haplophyllum hispanicum was found to possess cytotoxic properties and is twice as potent as 6-mercaptopurine, a well known cytotoxic agent². Anticancer screening tests with whole plant extracts of C. collinus on human epidermoid carcinoma of the nasopharynx in tissue culture, Walker carcinosarcoma 256 in the rat and L-1210 lymphoid leukaemia in the mouse have confirmed the anticancer activity of the plant. Cleistanthin A, another lactone glycoside of diphyllin, was found to possess remarkable neutrophilic granulocystosis activity in the rat, mouse, cat and monkey during drug-induced granulocytopenia^{10,11}. The biological activity of other compounds, namely cleistanthin B and collinusin, are under study,

Due to the toxic nature of the plant, clinicians and forensic toxicologists often encounter the problem of determining the active constituents in body fluids and tissues. A simple and at the same time precise protocol is therefore essential. The method reported here meets these requirements.

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